

VESELOVSKIY, I. N.

Tekhnicheskaya mekhanika; elementarnyi uchebnik dlia samoobrazovaniia. Moskva, Gostekhizdat, 1943. 282 p. diags.

Applied mechanics; elementary manual for self instruction.

DLC: TA350.V4

SO: Manufacturing and Mechanical Engineering in the Soviet Union, Library of Congress, 1953.

VESELOVSKIY, IVAN NIKOLAEVICH

Kurs mekhaniki dlia tekhnikumov. Dop. v kachestve uchebn. posobiia dlia tekhnikumov.  
Moskva, Gostekhizdat, 1947. 592 p. diagrs.

Course in mechanics for technical schools.

DLC: QA805.V3

SO: Manufacturing and Mechanical Engineering in the Soviet Union, Library  
of Congress, 1953.

VESELOVSKIY, I. N.

VESELOVSKIY, I. N. - "Babylonian Mathematics." Sub 26 Mar 52, Moscow Order of Lenin State U imeni M. V. Lomonosov. (Dissertation for the Degree of Doctor in Physicomathematical Sciences).

SO: Vechernaya Moskva January-December 1952

VESELOVSKII, I. N.

Tekhnicheskaya mekhanika; elementarnyi uchebnik dlia samobrazovaniia. Moskva,  
Gostekhizdat, 1943. 282 p. diagrs.

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Course in mechanics for technical schools.

DLC: QA805.V3

SO: Manufacturing and Mechanical Engineering in the Soviet Union, Library of Congress, 1953.

GERNET, Mikhail Mikhaylovich, prof.; SVESHNIKOV, G.N., zasl.  
deyatel' nauki prof., retsenzent; VESELOVSKIY, I.N.,  
doktor fiz.-mat. nauk, prof., retsenzent; POGOSOV, G.S.,  
kand. fiz.-matem. nauk, dots., nauchn. red.

[Course in theoretical mechanics] Kurs teoreticheskoi me-  
khaniki. Moskva, Vysshaia shkola, 1965. 406 p.  
(MIRA 18:7)

1. Moskovskoye vyssheye tekhnicheskoye uchilishche im. N.Ye.  
Baumana (for Veselovskiy).

MIKHALOVSKIY, A.G., doktor sel'skokhozyaystvennykh nauk; KALINERDA, V.I.,  
kand.biol.nauk; VESELOVSKIY, I.V., kand.biol.nauk.

Experience gained from grassland crop rotations practiced in the  
Ukrainian Polesye. Zemledelia 6 no.9:35-39 S '58. (MIRA 11:9)  
(Polesye--Rotation of crops)

MIKHALOVSKIY, A.G., doktor sel'skokhozyaystvennykh nauk, prof.; KALIBERDA, V.M., assistant; YAVORSKIY, A.G., kand.sel'skokhozyaystvennykh nauk, dotsent; VESELOVSKIY, I.V., kand.biologicheskikh nauk

Productivity of grassland crop rotations and measures for increasing soil fertility in the Ukrainian Polesye. Nauch. trudy UASHN 10:3-16 '60. (MIRA 14:3)

(Polesye- Rotation of crops) (Soil fertility)



VESELOVSKIY, I. V.

"The Effect of Perennial Grasses on Fertile Gray Forest Soils and Podsolized Chernozems Under the Conditions Which Exist in the Western Forest Steppes of the Ukrainian SSR."  
Cand Biol Sci, Kiev State U imeni T. G. Shevchenko, Kiev-L'vov, 1954. (KL, No 1, Jan 55)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (13)

SO: Sum. No. 598, 29 Jul 55

VESELOVSKIY, K. S.

O klimate Rossii (The Climate of Russia,) 9Pb, 1857.

S/138/62/000/012/007/010  
A051/A126

AUTHORS: Gamburg, D. Yu., Kazakov, A. V., Lelyakina, T. M., Belugina, L. N.,  
Veselovskiy, K. B.

TITLE: Investigation of carbon black produced by electro-cracking of  
natural gas to acetylene

PERIODICAL: Kauchuk i rezina, no. 12, 1962, 22 - 24

TEXT: Samples of acetylene carbon blacks, obtained from dry collection  
and produced in one of the electro-cracking plants, were studied in 1959 - 1960  
by the ГИАП (GIAP - State Institute of Scientific Research and Design of the  
Nitrogen Industry and Products of Organic Synthesis), in cooperation with  
НИИРП (NIIRP - Scientific Research Institute of the Rubber Industry). Inves-  
tigations were conducted to determine the possible use of these samples as fil-  
lers in rubber mixes. The major disadvantages of the investigated carbon blacks  
were found to be: the high volumetric numbers, elevated ash content and a low  
density which in some cases not exceeded 40 - 50 g/l. Work has been carried out  
to increase the density by 3 to 4 times and reduce the volumetric number from 34

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Investigation of carbon black...

8/138/62/000/012/007/010

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to 5.9 cm<sup>3</sup>/g. The ash content could also be reduced by regulating the production process through gas annealing with vapour condensate. Finally, the elevated content of volatile substances could also be reduced with an increase in annealing temperature. The advantages of the methane electro-cracking carbon black are: the high tensile strength, hardness according to TM-2 (TM-2), increased tear resistance exceeding the standard acetylene carbon black in this respect. It was experimentally established that with the properly adjusted carbon-black production process from gases of methane electro-cracking, carbon black compression, and its granulation, a stable product is formed which is not inferior to standard acetylene carbon black [ П-1250 (P-1250)], and carbon black from methane electro-cracking produced at present in the GFR. The investigated carbon black gives the same properties to the rubber mixes as the latter two. There are 2 tables.

ASSOCIATION: Gosudarstvennyy nauchno-issledovatel'skiy proyektnyy institut azotnoy promyshlennosti i produktov organicheskogo sinteza i Nauchno-issledovatel'skiy institut rezinovoy promyshlennosti (State Institute of Scientific Research and Design of the Nitrogen Industry and Products of Organic Synthesis and Scientific Research Institute of the Rubber Industry)

Card 2/2

VESELOVSKIY, Ivan Vasil'yevich[Veselovs'kyi, I.V.], dots.;  
MIKHALOVSKIY, A.G.[Mykhalovs'kyi, A.H.], prof., red.;  
YEFRIMOV, M.V., red.; VERNIK, G.V.[Vernik, H.V.], tekhn.  
red.

[Herbicides and their use in agriculture] Harbitsydy ta ikh  
zastosuvannia v sil's'komu hospodarstvi. Kiev, Derzhsil'-  
hospvydav URSS, 1964. 50 p. (MIRA 17:3)

GAMBURG, D.Yu.; KAZAKOV, A.V.; LELYAKINA, T.M.; BELUGINA, L.N.;  
VESELOVSKIY, K.B.

Investigating the carbon black obtained in the electric  
cracking of natural gas prior to the formation of acetylene.  
(MIRA 16:1)  
Kauch.i rez. 21 no.12:22-24 D '62.

1. Gosudarstvennyy nauchno-issledovatel'skiy proyektnyy  
institut azotnoy promyshlennosti i produktov organicheskogo  
sinteza i Nauchno-issledovatel'skiy institut rezinovoy pro-  
myshlennosti. (Carbon black) (Gas, Natural)

44162

S/181/62/004/010/062/063

B102/B104

24,7800

AUTHORS: Veselovskiy, P. F., and Suchkov, Yu. D.

TITLE: General case of resonator method of determining the dielectric constant

PERIODICAL: Fizika tverdogo tela, v. 4, no. 10, 1962, 2989-2992

TEXT: The theoretical bases are stated for a method of determining  $\epsilon$  in a cylindrical resonator that contains  $n$  sections filled with substances of different  $\epsilon$  (Fig. 1). For simplicity, energy dispersion is ignored and  $\mu=1$  over the whole volume. The mathematical solution of the problem is reduced to a considering the harmonic oscillations of the type  $H_{pqr}$  of the generalized resonator when the field components are

$$E_z=0, E_{z,z}=-\frac{\omega}{c} Z \nabla S,$$

$$H_z=r^2 Z S, H_{z,z}=\frac{dZ}{dz} \nabla S,$$

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General case of resonator method of ...

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$S=S(x_2, x_3)$  and  $Z=s(z) = A \sin(\alpha z + \varphi)$ .  $sZ=u(s, t_2, x_3)$  is the solution of the wave equation  $\Delta u + k^2 u = 0$  where  $k^2 = (\omega/c)^2 \epsilon = \chi^2 + \alpha^2$ . The boundary-value problem, together with the continuity condition, yields a system of equations of the form

$$\frac{\tan(\alpha_1 a_1)}{\alpha_1} = \frac{\tan(\alpha_2 a_1 + u_2)}{\alpha_2}, \dots, \dots, \dots = \frac{\tan(\alpha_n a_n)}{\alpha_n}$$

with the non-trivial solution

$$\begin{aligned} & \sum_{i=1}^n B_i - \sum_{i=2}^{n-1} \alpha_i^2 B_i \left( \sum_{j=1}^{i-1} B_j \right) \left( \sum_{k=i+1}^n B_k \right) + \sum_{\substack{i=2 \\ k \geq i+2}}^{n-1} \alpha_i^2 \alpha_k^2 B_i B_k \left( \sum_{j=1}^{i-1} B_j \right) \left( \sum_{l=i+1}^{k-1} B_l \right) \times \\ & \times \left( \sum_{m=k+1}^n B_m \right) - \sum_{\substack{i=2 \\ k \geq i+2 \\ l \geq k+2}}^{n-1} \alpha_i^2 \alpha_k^2 \alpha_l^2 B_i B_k B_l \left( \sum_{j=1}^{i-1} B_j \right) \left( \sum_{m=i+1}^{k-1} B_m \right) \left( \sum_{n=k+1}^{l-1} B_n \right) \times \\ & \times \left( \sum_{p=l+1}^n B_p \right) + \dots = 0. \end{aligned} \quad (2)$$

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$$B_i = \frac{\tan(\alpha_i a_i)}{\alpha_i}.$$



General case of resonator method of...

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This relation is the resonance condition for the magnetic oscillations and yields the parameters  $\alpha_1$  for determining  $\epsilon_1 = (c/\omega)^2(\alpha_1^2 + \kappa^2)$ .

$a = \sum_{i=1}^n a_i$  is the length and  $\omega_0$  is the resonance frequency of the resonator when  $\epsilon_1 = \dots = \epsilon_n = 1$ . In this case  $\kappa^2 = (\omega_0/c)^2 - (r\pi/a)^2$ , (3). For a dielectric of thickness  $a_2$  upon a dielectric base of thickness  $a_3$ ,

$$\alpha_2^2 = \frac{1}{B_1 a_2} \left[ 1 + (B_1 + a_2) \frac{1 - \alpha_3^2 B_3 B_4}{B_3 + B_4} \right]; \text{ if } \alpha_2^2 \text{ is put into Eq. (3) the}$$

dielectric constant  $\epsilon_2$  of the film can be determined. There are 2 figures.

ASSOCIATION: Leningradskiy politekhnicheskii institut im. M. I. Kalinina (Leningrad Polytechnic Institute imeni M. I. Kalinin)

Card 3/4

GANBURG, D.Yu.; LELYAKINA, T.M.; VESELOVSKIY, K.B.; BELUGINA, L.N.

Changes in the peat surface during its gasification. Inzh.-fiz.  
zhur. no.9:99-102 S '60. (MIRA 13:9)

1. Gosudarstvennyy institut azotnoy promyshlennosti, Moskva.  
(Peat gasification)

VESELOVSKIY, L. F.

U S S R

12307\* Investigation of the Relaxation Processes in Polyvinyl  
Acetate at Temperatures Below the Softening Temperature  
Исследование релаксационных процессов в поливинилацетате  
при температурах ниже температуры размягчения (Рус-  
сия) L. F. Veselovskiy and A. I. Shcheglov. Zh. obshch.  
khim. 25, no. 5, May 1955, p. 939-942.

Experimental investigations of the dielectric properties of a  
polymer. Measurements carried out in a temperature  
range of -150 to +20 C. and a frequency range of 50 to 10<sup>6</sup>  
cycles. Graphs to ref.

L 16468-66 EWT(m)/ETC(f)/EPF(n)-2/EWG(m) DM  
ACC NR: AP6005540 (N) SOURCE CODE: UR/0089/66/020/001/0075/0076

AUTHOR: Veselovskiy, L. N.; Kuznetsov, V. G.; Sakovich, V. A.

44  
B

ORG: none

TITLE: Optimum ratio of neutron- and gamma-radiation doses behind the shield of a reactor 19, 55

SOURCE: Atomnaya energiya, v. 20, no. 1, 1966, 75-76

TOPIC TAGS: radiation shielding, gamma radiation, neutron radiation, nuclear engineering, reactor shielding

ABSTRACT: It is shown that slight deviations from equality between the surface areas of the light and heavy components in a lead-water shield may have a considerable effect on the ratio of neutron- and gamma-radiation doses for optimum thicknesses of the water and lead components. No definite ratio of neutron- and gamma-radiation doses can serve as a generalized optimizing test depending on specific structural considerations. Therefore other tests must be used for checking optimum shielding conditions. Orig. art. has: 5 formulas.

SUB CODE: 18/

SUBM DATE: 11Mar65/

ORIG REF: 002/

OTH REF: 002

UDC: 621.039.58:539.125.5 + 539.122

2

Card 1/1 mc

ACC NR: AT6036520

SOURCE CODE: UR/0000/66/000/000/0099/0099

AUTHOR: Vesolovskiy, L. N.; Gribov, B. S.; Kuznetsov, V. G.; Sakovich, V. A.

ORG: none

TITLE: Measurement of absorbed doses of intermediate neutrons [Paper presented at the Conference on Problems of Space Medicine held in Moscow from 24 to 27 May 1966.]

SOURCE: Konferentsiya po problemam kosmicheskoy meditsiny, 1966. Problemy kosmicheskoy meditsiny. (Problems of space medicine); materialy konferentsii, Moscow, 1966, 99

TOPIC TAGS: neutron radiation biologic effect, cosmic radiation biologic effect, radiation shielding, radiation protection, radiation dosimetry

ABSTRACT: Study of the effectiveness of biological shielding of a nuclear reactor showed that the most convenient method of detecting intermediate-energy neutrons is neutron detection with preliminary moderation. The sensitivity of such detectors depends on moderator thickness, and also on the geometry of the moderator-detector system as a whole. Detectors with isotropic sensitivity received the most attention. In order to study the angular characteristics of neutron fluxes, a directional neutron detector with variable moderator thickness was created for biological shielding. The sensitivity of the detector was investigated with monoenergetic neutrons in the range 30 kev to

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ACC NR: AT6036520

15 Mev. It was found that use of different moderator thicknesses permits measurement both of neutron fluxes in the energy range 30 kev-18 Mev, and of the physical and biological doses produced by them. /W. A. No. 22; ATD Report 66-116/

SUB CODE: 06 / SUBM DATE: 00May66

Card 2/2

30 260

Volumetric zinc uranyl determination of sodium by the method of  
Dobbin and Byrd. N. Vesslovski (*Hydrochem. Acta*, 1941, 12, 25 --  
33). - [Na] of 11 samples of ground  $H_2O$  of Lower than district is  
given (cf. C., 1944, Part I). J. J. H.

VESELOVSKII, N.

P. KASHINSKII, H<sub>2</sub>Ochem. Material 7, 3-123, 1931

36

C-3

369. Continued to show exact determination of sodium in hydro-  
 chloric solution. *Hydrochem. Met.* 1941, 12,  
 1-54. 1 ml. of solution containing 0.2-0.5 mg. of Na is pptd.  
 with 10-15 ml. of  $\text{H}_2\text{UO}_4$  solution. *Barber and Kotched, A.,*  
 1940, 593, the ppt. washed with  $\text{H}_2\text{UO}_4$  saturated with Na Zn  $\text{UO}_4$ ,  
 acetate, with  $\text{H}_2\text{SO}_4$  and  $\text{H}_2\text{O}$ , and dried at 60-70°. Conditions  
 of washing and the effects of temp. and of the presence of  $\text{HCl}$ ,  
 $\text{H}_2\text{SO}_4$ ,  $\text{CaSO}_4$ ,  $\text{CaCO}_3$ ,  $\text{PO}_4^{3-}$ ,  $\text{SO}_4^{2-}$ , and humus are investigated.  
 The method gives for [Na] of lake and ground  $\text{H}_2\text{O}$  val. which  
 are higher by 1-3% than those found by pptn. with  $\text{H}_2\text{PtCl}_6$ .  
 I. J. B.

ASS-51A METALLURGICAL LITERATURE CLASSIFICATION

FROM SOURCE

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18

Solar heating of the mud from the Turolov river, and some of its other properties. N. Veshkovskii and M. Konarev. *Hydrachem. Material.* (U. S. S. R.) 10, 183-212 (in German 213) (1938). — The mud contains much org. matter, but little  $H_2O$ -sol. mineral material. It is not heated by the sun, since evapn. cools it. Ateln. of 120-30 g. NaCl per kg. of mud reduces evapn. and permits the mud to be heated by solar radiation, so that it can be used for medicinal purposes. H. M. Leicester

COMMON ELEMENTS										COMMON CHARACTERISTICS									
MATERIALS INDEX										PROCESS AND PROPERTIES INDEX									
<p>Changes in the water-soluble mineral parts of mud when it is kept or prepared for analysis. N. Veshnyak and M. Kovalev. <i>Hydrochem. Material.</i> (U. S. S. R.) 10, 218-227 (in German 238) (1968). - When river mud is kept for a long time, or stirred in air, sulfide S is oxidized to SO, and the total amt. of bound CO<sub>2</sub> is decreased. These changes are minimized if the sample is kept at a low temp. H. M. Leicester</p>										18									
<p>ASB-55A METALLURGICAL LITERATURE CLASSIFICATION</p>																			
FROM SOURCE										FROM SOURCE									
1-2										1-2									

VESELOVSKII, Nikolai Ivanovich

VESELOVSKII, Nikolai Ivanovich. ...Ocherk istoriko-geograficheskikh sviedeni  
o Khivinskom khanstvie ot drevnieishikh vremen do nastoiashchago. S.-Peterburg,  
1877. 364 p.  
OC1

SO: LC, Soviet Geography, Part II, 1951, Unclassified

*Veselovskiy, N. N.*  
VESELOVSKIY, N. N. and V. PIATON.

Aeros" emka gorodov. Moskva, Gosaviaavtoizdat, 1932. 168 p., illus.  
Bibliography: p.4.  
Title tr.: Aerial mapping of cities.

TR810.V4

SO: Aeronautical Sciences and Aviation in the Soviet Union, Library of  
Congress, 1955.

*Veselovskiy, N. N.*

VESELOVSKIY, N. N.

Fotogrammetriia. Dopushcheno...v kachestve uchebnogo posobiia dlia geodezicheskikh vuzov. Moskva, Izd-vo geodezicheskoi i kartograficheskoi lit-ry, 1945. 432 p., illus.

Bibliography: p. 431-432.

Title tr.: Photogrammetry. Approved as a textbook for institutes of advanced geodetic studies.

TA593.V45

SO: Aeronautical Sciences and Aviation in the Soviet Union, Library of Congress, 1955.

SHERSHEN', A.I.; VESELOVSKIY, N.N., redaktor; SHLENSKIY, I.A., tekhnicheskii redaktor

[Aerial photographic surveying; mapping process] Aerofotos'emka; letnos'emochnyi protsess. Moskva, Izd-vo geodezicheskoi i kartograficheskoi lit-ry, 1949. 251 p. (MLRA 9:11)  
(Aerial photogrammetry)



VESELOVSKIY, N.N., kandidat tekhnicheskikh nauk, dotsent.

Determining true angles of the inclination of photographs for measuring differences of horizontal parallaxes. Trudy MIIGAIK no.21:21-25 '55.  
(MIRA 10:1)

1. Moskovskiy institut inzhenerov geodesii, Kafedra fotogrammetrii.  
(Aerial photogrammetry)

PHASE I BOOK EXPLOITATION

808

*VESOLOVSKIY, Nikolay Nikolayevich*  
Vesolovski, Nikolay Nikolayevich

Aerofototopografiya (Aerial Phototopography) Moscow, Geodezizdat, 1958. 346 p.  
5,000 copies printed.

Ed.: Gebgart, Ya. I.; Tech. Ed.: Romanova, V. V.; Ed. of Publishing House:  
Khromchenko, F. I.

PURPOSE: This is a textbook in aerial phototopography for cartographic faculties  
of geodetic institutes.

COVERAGE: The book surveys the development of aerial phototopography and the part  
it plays in the national economy and especially in mapping of the country.  
Chapters on linear perspective, analysis and interpretation of aerial photographs,  
sketching-in of relief, and the universal and differential methods of surveying  
are included. Positional and height condensations of points and methods of  
making original maps are also available. The book was reviewed by the following  
two Soviet scientists, both of the Moscow Institute of Geodesy, Cartography and  
Aerophotography: M. D. Konshin, Doctor of Technical Sciences, and Ya. I.  
Gebgart, Candidate of Technical Sciences. There are 232 figures and 52 ref-  
erences, of which 44 are Soviet, 6 English, 1 German and 1 French.

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# Aerial Phototopography

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MM/flc  
11-24-58

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. '60.

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geodezii, aerofotos"yemki i kartografii.  
(Aerial photogrammetry)

807/6-59-6-21/22

Chralole (Dronjka)

Ододану и картографи, 1959, № 6, стр. 74-75 (УРСН)

[illegible][illegible]

Case 2/4

Card 3/4



CA

17

Scheme for analysis of medicinal mud and the form of recording its results. P. A. Kashinskii and N. V. Yezhovskii. *Gidrotekhn. Materialy (Hydrochem. Materials)* 13, 3-21 (in English, 31-2)(1947).—The merit of the Shchukarev method for analysis of medicinal mud (C.A. 25, 57) was emphasized. However, the following inadequacies of the scheme were pointed out: (1) the amts. of "decompos. products of siliceous particles" are low by 30-70% or more, because silicic acid nepsd. as a gel is not included; (2) results of mech. analysis are too high because of silicic acid rel; and (3) the contents of the adsorbed cations given are unreliable. Results obtained by different investigators using the Shchukarev scheme showed disagreements in the following details: (1) moisture, (2) gypsum, (3) siliceous particles (those larger than 1  $\mu$  in diam. as well as smaller ones), (4) basic oxides and adsorbed cations that are included in the composition of the colloid complex. Results of analyses obtained by several investigators using the Shchukarev method are provided. A modified scheme of analysis, based on the Shchukarev method, is presented. (Nadys S. Macy)

CA

16

Preparation of a mud solution by pressure or suction.  
~~N. V. A. Kiselevskiy and I. I. Kiselevskiy.~~ *Materials* (Hydrochem.  
 Materials) 11, 21-31 (1947).—For 3 samples of medicinal  
 mud with differing compn. of the liquid phase, exts. of  
 the same muds prepd. by pressure or by suction analyzed  
 practically the same as their resp. liquid phases. Among  
 the advantages of the suction method are: a transparent  
 mud ext. is obtained, the danger of oxidation, contamina-  
 tion, etc., is reduced, and one vacuum pump may be used  
 in prepn. of several mud exts. For each of the two methods  
 of prepn., the following results are listed: d. of ext.; pH;  
 wt. % of  $\text{CO}_3$ ,  $\text{Cl}^-$ ,  $\text{SO}_4^{--}$ ,  $\text{S}_2\text{O}_3^{--}$ ,  $\text{HS}^-$ ,  $\text{CO}_3^{--}$ ,  $\text{HCO}_3^-$ ,  
 $\text{Mg}^{++}$ ,  $\text{Ca}^{++}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ; and wt. of dry residue and amt. of  
 org. material. Gladys S. Macy

CA

9

Water-soluble portion of mud from the data of analysis of mud solutions and aqueous extracts. N. V. Veselivskii. *Geokhimiya. Materialy (Hydrokhim. Materialy)* 13, 31-32 (English summary, 62(1947)).—In aq. exts. of mud prep'd. without washing, the content of all ingredients was, on an av., 3.15 times higher than for solns., and in exts. prep'd. with washing from chlorides it was 12.4 times higher. Alteration of the water-sol. portion of the mud during prep'n. of exts. is caused by soln. of the solid phase of the mud, by metathesis of the cations of the soln. and the colloid complex, and by oxidation of the sulfide S. With reference to these processes the analyzed muds can be divided into the following groups: (1) highly saline muds contg. gypsum in the solid phase and (2) wet muds of decreased salinity and viscous mud almost devoid of salts. In ext. prep'd. from the first type of mud the soln. of material from its solid phase and metathesis of Ca from gypsum with adsorbed cations have the greatest effect on alteration of the compn. of the water-sol. part of the mud. When exts. are prep'd. from the 2nd type of mud there is oxidation of sulfide S by atm. O, and this oxidation is the more intense the less is the content of salts in the mud. When aq. exts. prep'd. without washing are used to establish the compn. of the water-sol. portion of the mud, they should be prep'd. with a const. solid to liquid ratio. Many tables of chem. analyses of different mud samples are provided.

G. L. S. Macy



CA

17

Computation of the results of analysis of aqueous extracts of medicinal muds. N. V. Veselovskii. *Gidrobiokh. Materialy (Hydrochem. Materials)* 13, 53-7(1947)(English summary); cf. C.A. 45, 6134c.—It was proposed that results of analyses of aq. exts. of medicinal muds be expressed in g. per 100 g. of dry material of the mud as the amts. of substances per total amt. of water added in prepn. of the ext. and that present in the crude mud. Thus, in the case of saliferous muds, it is possible to avoid an error caused by disregarding the portion of the vol. of the ext. occupied by the dissolved salts. It was found that the percentage of moisture in the mud, estd. by drying with  $\text{Na}_2\text{CO}_3$  at  $145^\circ$ , was evidently exaggerated by 1 or 2 percent.

Gladys S. Macy

CA

Concentration and volume of hydrochloric acid necessary for the decomposition of carbonates of medicinal mud in preparation of the extracts. P. A. Kashinskii and N. V. Verkhovskii. *Gidrokhim. Materialy (Hydrochem. Materials)* 13, 68-70 (1947).--The following conclusions resulted from a study of 13 samples of medicinal muds of different origins and compns.: (1) the carbonates of the muds are decompd. by boiling for 30 min. with enough HCl so that, besides the quantity calcd. to react with sulfides and carbonates of the mud, there are at least 10 ml. of 0.2 N HCl/g. of dry material and (2) sep. exts. should be made for detn. of gypsum in muds with high content of the latter. The data on which these conclusions are based are included. G. S. Macy

CA

Extraction by alkali of silicic acid, the product of the decomposition of silicates of mud by hydrochloric acid. N. Y. Venzlovskij. *Gidrokhim. Materialy (Hydrochem. Materialy)* 13, 81-83(1947) (English summary).—Tests showed that 0.5% NaOH soln. cannot be used in analysis of medicinal muds. In mud residues contg. large amts. of calcium sulfates no silicic acid could be detected by using 0.5% NaOH soln. About 1.5% of that acid was found by the sodium carbonate method. Calcined mud residues gave higher silicic acid contents than uncalcined ones because of decompn. of kaolin with liberation of silicic acid during calcining. Tabulated data comparing the NaOH method with the  $\text{Na}_2\text{CO}_3$  method were provided. Gladys S. Macy

CA

14

Muds of Tuzlova River. N. V. Yezlovskii. *Gidrobiol. Materialy (Hydrobiol. Materials)* 13, 91-107 (English summary, 107 88(1947); cf. C.I. 32, 6012). Study of the medicinal muds of Tuzlova River near Novotcherkassk was made in 1941. Within this territory were discovered 4000 to 5000 tons of dark gray mud suitable for mud-baths after wash. of NaCl. Samples collected over a range of 10 km. were found to be similar in chem. compn., but greatly variable as to retention of water and mech. compn. On an av. 100 g. of mud, dried at 145° contained the following: (1) 10.6 g. org. matter, (2) 4.3 g.  $\text{CaCO}_3$ , (3) 0.57 g. iron sulfide, (4) 85.1 g. of silicates together with quartz sand, and (5) 0.5 g. water-sol. substance. It was established that the higher the content in the mud of particles of less than 10  $\mu$  diam., the greater the amt. of water that could be retained. Mech. analysis of mud should be restricted to particles with 250 to 60  $\mu$  diam., 5 to 10  $\mu$ , and  $\approx$  10  $\mu$ . There are tables giving the results of chem. and mech. analyses. G. S. M.

CA

Determination of silica in hydrochloric acid extract of medicinal mud. N. A. Vostorzhnik (Hydrochem. Inst., Novocherkassk). *Gidrokhim. Materialy* (Hydrochem. Materials) 14, 31-4 (1948). -- Due to the presence of as much as 18% of gypsum,  $MgCl_2$  (as much as 8.8%), and other salts, the conventional  $SiO_2$  detn. gives higher figures. To overcome this, the following method of analysis was developed. After oxidizing the org. matter with aqua regia, the HCl ext. was evapd. to dryness and the residue dried for 1 hr. at  $110^\circ$ . To this, 37% HCl was added, mixed with the residue, and  $H_2O$  added to make a 6% HCl soln. Enough of this acid was added to keep the gypsum and other salts in soln. After 1 hr. on the water bath the  $SiO_2$  was filtered. If the gypsum persisted, more 8-10% HCl was added and heated for 15 min. and filtered. The  $SiO_2$  was washed with 1% hot HCl until no test for Fe was obtained. The  $SiO_2$  was treated with HF and purity detd. by the residue left behind. The filtrates were evapd. and dried at  $110^\circ$  several times until no more  $SiO_2$  sepd. J. S. Joffe

CA

Determination of ferrous iron in medicinal mud. N. V. Veselovskii (Hydrochem. Inst., Novocherkassk). *Gidrometallurgiya* (Hydrochem. Materials) 14, 35-7 (1948).  
For detg. Fe<sup>++</sup> in medicinal mud, the method of Kashinskii and Slavskii (combining the H<sub>2</sub>S and HgCl<sub>2</sub>) (Separate Bull., Leningrad, 1931) was found to be better than that of E. S. Burkner and V. V. Burkner (*C.A.* 34, 5198). J. S. J.

CA

The silicate portion of the medicinal mud according to hydrochloric acid extracts and total analyses. N. V.

~~Yegorovskii~~ Hydrochem. Inst., Novocherkassk). *Gidrokhem. Materialy* (Hydrochem. Materials) 15, 40 (1948).  
The silicates of the mud are decomposed least when 0.7% HCl as compared with 5 and 10% HCl is used in the extraction. The constituents affected most by the HCl treatment are Mg, Fe, and  $\text{SiO}_2$ . The composition of 5 different muds is given.  
I. S. Joffe

CA

Determining the total organic matter of medicinal mud  
by loss on ignition. *N. V. Novotsvetov* (Hydrochem. Inst.,  
Novocherkassk). *Gidrokhim. Materialy* (Hydrochem. Ma-  
terials) 15, 69-71 (1948).—On igniting for total org. matter  
the loss of carbonates and chlorides is to be considered in  
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1. Hydrochemical characteristic of the ponds in the arid

6-20

that is, the amount of water in the ponds is determined on the basis of the results of the hydrochemical analysis of the water.

VESELOYSKIY, N. V.

✓ The effect of surface and subsoil feeding upon the regime  
of the main ions of the pond

Tarasov, M. V. and N. V. Veselovskiy and M. G. G.  
Cherkassia

The effect of surface and subsoil feeding upon the regime  
of the main ions of the pond

territory showed that the main influx consists of highly  
mineralized subsoil waters of the sulfate class. Hydrocar-  
bonated waters of low mineral content reached the pond dur-  
ing heavy snow melting and rainfalls. A. I. Murkin

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126, 2, p. illus., diags., maps, tables (Nauchno-populyarnaya  
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*Veselovskiy, N. V.*

Category: USSR

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Abs Jour: RZh--Kh, No 3, 1957, 7877

Author : Veselovskiy, N. V. and Goncharova, I. A.

Inst : Not given

Title : Establishment of and Variations in the Main Ion Composition in the Water of a Reservoir in Rostov Oblast

Orig Pub: Gidrokhim. Materialy, 1955, Vol 25, 115-153

Abstract: The results of a three-year (1951-1953) study of hydrochemical conditions in a reservoir located in the northwestern portion of the subnormal rainfall section of Rostov Oblast are reported. It has been found that in 1951 after the filling of the reservoir by surface run-off the concentration of the main ion species varied over the between-floods period (Alekin classification index of  $CCa$ ). The mineral content increased from 100 mg/liter after the spring floods to 400 mg/liter in the winter. During the flooding the ion composition is established as the result of the mixing of surface water flowing along the slopes and valley bottom, ground water,

Card : 1/2

-50-

Category: USSR

D

Abs Jour: RZh--Kh, No 3, 1957, 7877

and the water remaining in the reservoir at the onset of the flood season. The seasonal changes in the ion composition result from the seepage of subsurface water into the reservoir, the loss of water by filtration and evaporation, and chemical, biochemical, and biological processes taking place in the water of the reservoir. Of the total seasonal change in mineral content, evaporation accounts for 6.7-20.9% in separate years and subsurface water seepage, 79.1-93.3%. The loss of water by filtration from the reservoir between the spring flood and the formation of the first ice crust represents 36.2-44.2% of the spring water volume.

Card : 2/2

-51-

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(Serpukhov--Gas, Natural) (Pressure regulators)

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the European territory of the U.S.S.R. Trudy Lab. ozeroved. 7:129-133  
'58. (MIRA 11:10)

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(Farm ponds)

V ESELOVSKIY, N. V

USSR / Cosmochemistry. Geochemistry. Hydrochemistry.

D

Abs Jour : Ref Zhur - Khimiya, No 3, 1957, No 7877

Author : Vosolovskiy, N.V., and Goncharova, I.A.

Inst : Not given

Title : Establishment of and Variations in the Main Ion Composition in the Water of a Reservoir in Rostov Oblast.

Orig Pub : Gidrokhim, Materialy, 1955, Vol 25, 115-153

Abstract : The results of a three-year (1951-1953) study of hydrochemical conditions in a reservoir located in the northwestern portion of the subnormal rainfall section of Rostov Oblast are reported. It has been found that in 1951, after the filling of the reservoir by surface run-off, the concentration of the main ion species varied over the between-floods period (Molokan classification index of  $C^{0a}$ ). The mineral content increased from 100 mg/liter after the spring floods to 400 mg/liter in the winter. During the flooding, the

Card : 1/2



USSR / Cosmochemistry, Geochemistry, Hydrochemistry.

D

Abs Jour : Ref Zhur - Khimiya, No 3, 1957, No 7877

Abstract : Ion composition is established as the result of the mixing of surface water flowing along the slopes and valley bottom, ground water, and the water remaining in the reservoir at the onset of the flood season. The seasonal changes in the ion composition result from the seepage of subsurface water into the reservoir, the loss of water by filtration and evaporation, and chemical, biochemical, and biological processes taking place in the water of the reservoir. Of the total seasonal change in mineral content, evaporation accounts for 6.7 - 20.9% in separate years and subsurface water seepage, 79.1 - 93.3%. The loss of water by filtration from the reservoir between the spring flood and the formation of the first ice crust represents 36.2 - 44.2 % of the spring water volume.

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Regime of dissolved gases and biogenic substances as exemplified  
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(MIRA13:9)

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(Electric engineering--History)

VESELOVSKIY, O. N.

AID P = 453

Subject : USSR/Electricity  
Card 1/1 Pub. 27 - 16/34  
Author : Veselovskiy, O. N., Kand. of Tech. Sci., Moscow  
Title : The Role of M. O. Dolivo-Dobrovolskiy in the Development  
of the Theory of Alternating Current  
Periodical : Elektrichestvo, 7, 77-80, J1 1954  
Abstract : The works of Dolivo-Dobrovolskiy are described.  
2 diagrams, 9 references (1887-1948).  
Institution : None  
Submitted : No date

VESELOVSKIY, O.N., kandidat tekhnicheskikh nauk.

Work of M.O.Dolivo-Dobrovol'skii. Elektrichestvo no.1:78-82  
Ja '56. (MLBA 9:3)

1. Moskovskiy energeticheskiy institut imeni Molotova.  
(Dolivo-Dobrovol'skii, Mihail Osipovich, 1862-1919)

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CHILKIN, M.G.; MESHKOV, V.V.; GOLUBTSOVA, V.A.; SIROTINSKIY, L.I.; VENIKOV, V.A.;  
ZOLOTAREV, T.L.; KONFEDKRA TOV, I.Ya.; SHNEYBERG, Ya.A.; ~~VESLOVSKIY, O.N.~~

Professor L.D.Bel'kind. Elektrichestvo no.8:93-94 Ag '56. (MLRA 9:10)  
(Bel'kind, Lev Davidovich, 1896-)



VESELOVSKIY, O.N. (Novosibirsk)

M.O. Dolivo-Dobrovol'skii; on the 100th anniversary of his birth.  
Vop. ist. est. i tekhn. no.13:148-149 '62. (MIRA 16:5)

(Dolivo-Dobrovol'skii, Mikhail Osipovich, 1862-1919)

VESELOVSKIY, Oleg Nikolayevich; LEVIT, Ye.I., red.izd-va;  
GRIGOR'YEVA, Ye.I., tekhn. red.; LAUT, V.G., tekhn. red.

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(MIRA 16:4)

(Electric engineering)  
(Dolivo-Dobrovol'skii, Mikhail Osipovich, 1862-1919)

VESELOVSKIY, O.H., kand.tekhn.nauk; KONFEDERATOV, I.Ya., doktor tekhn.nauk;  
SHNEYBERG, Ya.A., kand.tekhn.nauk

Prerequisites and importance of the development of electrical power  
engineering. Trudy MEI no.26:9-29 '57. (MIRA 11:9)  
(Electric engineering)

AUTHOR: Veselovskiy, O. N., Candidate of Technical Sciences SOV/105-58-9-16/34

TITLE: The Magnetic Rotating Field (Vrashchayushcheyesya magnitnoye pole) 70<sup>th</sup> Anniversary of Its Discovery (K 70-letiyu otkrytiya)

PERIODICAL: Elektrichestvo, 1958, Nr 9, pp 66 - 70 (USSR)

ABSTRACT: A brief historical survey is given here. The discovery of the magnetic rotating field made by G. Ferraris and the Yugoslavian Nikola Tesla, and their lives, are described. The works of Arago (1824), of Deprez (1883), and of some other physicists are mentioned. It is pointed out that Tesla made his discovery as early as 1882 while Ferraris followed in 1885. The German patent Nr 47885, and the British patent Nr 6481 that were granted to Tesla are briefly described. Finally it is stated that the Russian M.O.Dolivo-Dobrovolskiy in 1888 - 1889 had realized the essential error committed by Ferraris, and had developed all the features of the three-phase system which have been valid ever since in the same form. There

Card 1/2

*VESELOVSKIY, O.N.*  
 ALIKSANDROV, A.G., dots; ARONOVICH, I.S., inzh.; BABIKOV, M.A., doktor  
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 METUSHIL, A.V., doktor tekhn.nauk; NIKULIN, N.V., kand.tekhn.nauk;  
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 doktor tekhn.nauk; SOLOV'YEV, I.I., doktor tekhn.nauk; STUPEL' F.A.,  
 kand.tekhn.nauk; TALITSKIY, A.V., prof.; TEMNIKOV, F.Ye., kand.tekhn.  
 nauk; FEDOROV, L.I., inzh.; FEDOSEYEV, A.M., doktor tekhn.nauk;  
 KHOLYAVSKIY, G.B., inzh.; CHECHET, Yu.S., doktor tekhn.nauk; SHNEY-  
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[The history of power engineering in the U.S.S.R. in three volumes]  
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ALEKSANDROV, A.G.—(continued) Card 2.

Vol.2. [Electric engineering] Elektrotehnika. Avtorskii kollektiv  
toma: Aleksandrov i dr. 1957. 727 p. (MIRA 11:2)

1. Moscow. Moskovskiy energeticheskiy institut. 2. Chlen-korrespon-  
dent AN SSSR (for Larionov)  
(Electric engineering)

L 24807-66 EWT(m)/EPF(n)-2/ENP(j)/T/EWA(h)/ETC(m)-6/EWA(1) IJP(c) WW/GG/RM  
 ACC NR: AP6012722 (A) SOURCE CODE: UR/0190/66/008/004/0744/0748

AUTHOR: Veselovskiy, P. A.; Leshchenko, S. S.; Karpov, V. L.

ORG: Physicochemical Scientific-Research Institute im. L. Ya. Karpov (Fiziko-khimicheskiy institut) .

TITLE: Thermal degradation of irradiated polypropylene

SOURCE: Vysokomolekulyarnyye soedineniya, v. 8, no. 4, 1966, 744-748

TOPIC TAGS: pyrolysis, irradiation, polypropylene, molecular structure, chain polymer, gel

ABSTRACT: Changes in the structure of irradiated polypropylene have been studied by pyrolysis. Polypropylene chains were found to contain active groups which appear to be oxygen-containing groups of various structure. The increase in gas formation at the initial stage of pyrolysis for nonirradiated polypropylene is caused by the presence of the active oxygen-containing groups; and for the polypropylene, irradiated up to the gel-formation dose, it is caused by the presence of branching points in the chain. The active (oxygen-containing) groups are spent with the irradiation of polypropylene. Since the polypropylene chains irradiated below the dose for the initial stage of gel-formation have few branchings, a drop in the characteristic viscosity in polypropylene irradiated with small doses is caused mainly by degradation of the molecular chains. Cross-linking of the polypropylene chains is inhibited by

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L 24807-66

ACC NR: AP6012722

the presence of active (oxygen-containing) groups in them. Atactic polypropylene is found to be a stereoregular branched polymer. The rate of degradation of cross-linked polypropylene is higher than that of linear polypropylene. Orig. art. has: 6 figures and 4 formulas. [AM]

SUB CODE: 07/

SUPM DATE: 07May65/

ORIG REF: 005/

OTH REF: 010/

Card 2/2



VESELOVSKIY, P.F. [Veselova'kyi, P.F.]

Some advantages of the dielectric method in studying the structure and intermolecular forces using a polymer - solvent model in the liquid state. Ukr. fiz. zhur. 9 no.1:99-100 Ja '64.  
(MIRA 17:3)

1. Leningradskiy politekhnicheskoy institut im. Kalinina.

VESELOVSKIY, P. F.

"An Investigation of the Dielectric Properties of Polymers in the Centimeter Range of Radiowaves." Cand Phys-Math Sci, Leningrad Polytechnic Inst, Leningrad, 1953.  
Dissertation (Referativnyy Zhurnal--Fizika Moscow, Feb 54)

So: SUM 186, 19 Aug 1954

USSR/Physics - Dielectric losses

FD-3043

Card 1/2 Pub. 153 - 12/23

Author : Veselovskiy, P. F.

Title : Dependence of coefficient of dielectric losses epsilon (e") of polar polymers upon temperature

Periodical : Zhur. tekhn. fiz., 25, February 1955, 266-269

Abstract : In this work the author shows that the coefficient of dielectric losses epsilon (e" = e' . tan d) of polar polymers in the region of the maximum is a function of the absolute temperature T and distribution parameter of relaxation time alpha. He concludes that the empirical equation of Fuoss and Kirkwood can be applied to clarify the following frequently encountered experimental fact: the decrease or increase of the loss angle tangent (tan d) in the region of the maximum with variation of temperature or frequency of the variable electrical field, and that the value of the coefficient of dielectric losses at the maximum depends upon the ratio of alpha to T. Further, the distribution parameter of

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FD-3043

Abstract : relaxation time practically does not depend upon temperature in the case of dipole-radical dielectric losses, where the value of tangent of the loss angle  $\delta$  decreases at the maximum with increase of temperature and vice versa. Eight references.

Institution : -

Submitted : March 15, 1954

USSR/Physics - Dielectrics

FD-2824

Card 1/1      Pub. 153-7/30

Author      : Veselovskiy, P. F.

Title      : Measurement of  $\epsilon'$  and  $\tan \delta$  of a Solid Dielectric on Centimeter Radiowaves in a Temperature Range of  $-100$  to  $+100^\circ\text{C}$

Periodical      : Zhur Tekh. Fiz, 25, 601-609, 1955

Abstract      : A new method for measuring of  $\epsilon'$  and  $\tan \delta$  of a solid dielectric in a wide temperature range and on a wave length of 3.17 cm consisted in using a rectangular resonator designed for  $H_{10}$  oscillations and the method of contactless linking of waveguide sections. Results of tests are illustrated in graphs and tables and are in good agreement with the works of P. P. Kobeko, Kuvshinskiy, Shishkin and Mitsushim. Eleven USSR references and 5 foreign.

Institution      :

Submitted      : January 30, 1954

VESELOVSKIY, P. F.

2-11-61

517.224.33 : 621.315.616.06  
1201 INVESTIGATION OF THE RELAXATION PROCESSES  
IN POLYVINYL ACETATE AT TEMPERATURES BELOW THE  
SOFTENING TEMPERATURE. P. F. Veselovskii and  
A. I. Slutsker.

Zh. tekhn. Fiz., Vol. 25, No. 5, 939-42 (1955). In Russian.  
The investigation covered the temperature range (-150° to  
+20° C) and frequency range 50-10<sup>10</sup> c/s. At room tempera-  
ture tan δ has a low maximum which at lower temperatures  
shifts towards lower frequencies, indicating the relaxation  
character of the dielectric losses. Plastication and inter-  
linkage of the polar radicals hardly affect the frequency rela-  
tion of tan δ. Results seem to confirm Motek's hypothesis  
that the polarization losses in polar polymers below softening  
temperature are due to the thermal movements of the polar  
radicals. The polymer chains are practically immobilised.

Electrical Research Association

VESELOVSKIY, P.F.

2A. 0

✓ 1202 INVESTIGATION OF RELAXATION PROCESSES IN  
POLYVINYL ACETATE. P.F. Veselovskii and A.I. Gutsik.  
Zh. tekh. fiz. Vol. 25, No. 7, 1294-8 (1955). In Russian.  
The dielectric losses in polyvinyl acetate were measured  
in a wide temperature range above the softening temperature  
at the frequency 100 Hz. The experimental results were  
theoretically evaluated by considering a relation of the form  
 $\tan \delta = f(T)$  where  $f$  and  $T$  are frequency and temperature  
at which  $\tan \delta$  is a maximum. Comparison with results ob-  
tained for solid p.v.a. enabled the two components distinguished  
as dipolar-relaxation losses and dipolar-radical dielectric relaxa-  
tion losses to be separated.

Electrical Research Association

VESELOVSKIY, P.F. [Veselovs'kyi, P.F.]

Some characteristics of the occurrence of dipole relaxation  
in solutions of polymethylmethacrylate-toluol. Ukr. fiz.  
zhur. 9 no.10:1115-1121 0 '64 (MIRA 18:1)

1. Leningradskiy politekhnicheskii institut.



VESELOVSKIY, P.F.; MATVEYEV, V.K.

Dielectric properties of stereoregular polymethylmethacrylate  
polymers solutions in toluene. Vysokom. soed. 6 no.7:1221-1226  
Jl '64 (MIRA 18:2)

1. Leningradskiy politekhnicheskii institut imeni Kalinina.

VESELOVSKIY, P.F.

Nature of the dipole-radical relaxation of amorphous polymers in a solid  
vitreous state. Plast.massy no.12:59-60 '63. (MIRA 17:2)

VESELOVSKIY, P.F.; SUCHKOV, Yu.D.

Use of the resonance loop method in determining  $\tan \delta$  in dielectrics.  
Fiz. tver tela 5 no.9:2728-2730 S '63. (MIRA 16:10)

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VESELOVSKIY, P.F.; VOROB'YEVA, Ye.P.

Dielectric properties of styrene stereocopolymers. Plast.massy  
no.2:6-11 '63. (MIRA 16:2)

(Styrene polymers—Electric properties)  
(Butadiene)